Graphing Review

1-20 Using the following chart of an unknown liquid, make a graph. Consider volume as the independent variable. Be sure to include axis labels, a title, even divisions, and straight lines.

| | | | | | | | | | | | | Volume of liquid in mL Mass of liquid in g |
|---|---|---|--|--|--|-----|--|--|--|--|---|--|
| | | | | | | | | | | | | 20 18 |
| | | | | | | | | | | | | 100 90 |
| | | | | | | | | | | | | 75 67.5 |
| | | | | | | | | | | | | 40 36 |
| | | | | | | | | | | | | 10 9 |
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| | | | | | | | | | | | | 21. As mass increases, what happens |
| | | | | | | | | | | | | to the volume? INCREASES |
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| | | | | | | | | | | | | 22. As volume increases, what happens |
| | | | | | | | | | | | | to the mass? INCREASES |
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- 23. While looking at your graph, estimate how many grams 80 mL would come out to. 72g
- 24. How many grams would 50 mL come out to? 45g
- 25. How many grams would 60 mL come out to? 54g
- 26. What volume would 36g occupy? 40mL
- 27. What volume would 81g occupy? 90mL
- 28. What volume would 72g occupy? 80mL
- 29. What is the density of the liquid? (Show ALL work)

90/100 = .9 g/mL

30. Is this more or less dense than water, (which is 1 gram per ml)? LESS DENSE