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## STATION 1 – PEPSI VERSUS DIET PEPSI

Calculate the density of the pepsi can and the diet pepsi can. To find the volume of the pop, use information from the can. Please note that there is air in the top of every can that also has volume (takes up space). The air volume is approximately 23 mL. Add this to the liquid volume to get the total volume. Find the mass of each can using the triple beam balance. Record your data below IN A NEAT CHART. Include mass, volume, and density. (Put a unit label next to all numbers) (12 pts)

- A] Densities above what number sink in water?
- B] Objects or liquids with what density float in water?
- C] Which can will float? DO NOT GUESS OR EXPERIMENT! Prove that your answer is correct to the teacher BEFORE you confirm your hypothesis by testing!

## STATION 2 -METAL CYLINDER VERSUS METAL CYLINDER

Calculate the density of the long cylinder and the short cylinder. Use the water displacement method to find the volume. Find the mass of each cylinder using the triple beam balance. Record your data below IN A NEAT CHART. Include mass, volume, and density. (Put a unit label next to all numbers) (12 pts)

## **STATION 3- CUBE VERSUS CUBE**

Handle all the blocks in the black trays. DO NOT mix up the sets of blocks. Using *just your hands* and your *brains*, determine which block in the tray is more dense. Make a chart that shows your data. MAKE IT NEAT. (4pts)

## **Station 4- Mystery Minerals**

Find the density of the 2 minerals at this station. U	Jse
the chart to hypothesize which mineral is which.	
Make a chart that shows your data. MAKE IT NE	AT.
(12 pts)	

A] I think the hexagon mineral isB] I think the crystal mineral is	-
C] If gold and fool's gold look the same, how do yo tell the difference?	u

D] If you were to hold a cube of graphite and a cube of galena in each hand (both cubes having the same volume) Which would feel heavier to your hand?

Densities of some common minerals and metals,				
in grams/cubic centimeter				
	Density	Color description of		
		mineral		
ice	0.9	Clear		
sulfur	2.1	Yellow in color		
halite	2.2	Clear		
gypsum	2.3	Opaque white		
graphite	2.3	Silvery grey		
feldspar	2.6	Orange, pink, or green		
calcite	2.7	Clear		
quartz	2.7	Clear		
fluorite	3.2	Clear green, purple, blue		
corundum	4.0	Brown, blue, green		
Pyrite	5.0	Gold color		
(fools gold)				
magnetite	5.2	Grey to black		
galena	7.5	Silvery grey		
copper	8.9	Orange metallic		
native silver	10.5	White metallic		
gold	19.3	Gold color		

E]	Draw the	atoms i	n a cube of	F silver and a cube of gold. (2pts)
	silver		gold	

F] What is the definition of density? NOT THE FORMULA!! What does it mean?